# Computer Simulation of Precise Dew Point Initiation in Gas Pipelines Network System

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**ABSTRACT:-** Mathematical models have been developed to ascertain the exact position of commencement of dew point in gas pipelines network system transmitting gaseous components or complex hydrocarbon mixtures. Attainment of dew point in gas transmission pipelines leads to flow separation. Inadvertently, two-phase flow situation will result with liquid and gas holdups coexisting. The liquid holdup eventually settles at low points in the pipe as condensates. The condensate so formed obstructs flow leading to increased pressure drop along the line. The driving force of a flowing fluid stream is a function of the overall line pressure drop and line throughput. The higher the two militating factor, the higher the pumping and compression power. It is very worthy to note the exact point on a gas pipeline at which condensation will set in; drain plugs could be installed at such points for periodical removal of the condensate so formed. This will offset the high cost of investment on gas pipelines assets facilities.

**KEYWORDS:-** Burial Depth, Dew Point Temperature, Gas Fraction Mass flow-rate, Specific Heat Capacity, Internal fluid Convective Heat Transfer, Thermal Conductivity, Mixture Velocity, Heat Flux, Residual Heat, Flow Resident Time, Heat Flux.

Date of Submission: 18-04-2018

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Date of acceptance: 03-05-2018

### **I** INTRODUCTION

Gas pipelines network system run over a distance of thousands of kilometres traversing places of diverse temperature gradients [1, 2, 3]. This necessitates the need to find at what point along a pipeline at which condensation will commence after attainment of dew point for any particular gas or natural gas gaseous mixtures. The computer simulation approach is geared toward the determination the dew point temperature and the exact point or distance on the pipeline network system where the dew point of the hydrocarbon mixture will set in. At such points the remedial measure that could be taken is the instalment of drain plugs for removal of condensates; thereby getting rid of the problem of increased pressure drops that will ultimately result in increased pumping or compression power. This approach will result in a reduction of capital cost of pipeline assets and facilities even the inline equipment.

Hydrocarbon dew point determination is it the gas chromatography method, Chilled mirror method, or gas equation approach still shows a wide margin of deviation when results are compared. The dew point temperature is highly dependent on the operating steam pressure, environmental effects and gas composition [4, 5, 6]. In general terms, the accuracy of these methods is within -5% to 5%. Hence the need to consistently formulate models that could correlate the experimental results and mathematical models results to a close margin of accuracy.

# II PURPOSE AND SIGNIFICANCE

Gas pipelines systems with reference to Natural Gas Pipelines approach due points temperatures by the reason of environmental effects, average flow stream pressure, bulk stream temperature, cooling associated with pressure drops at valves and fittings, the sudden expansion of the fluid stream in the pipeline among others. These effects can lead to condensation or liquid fallout from heavy hydrocarbon components of the gaseous mixture. The liquid fallout if not timely recovered could lead to increased line pressure drop, decreased the heating value of gas stream, increased pumping and compression power, reduction in line transmission efficiency, higher cost of investment and operation of gas pipeline assets and facilities and other undesirable effects.

In the gas pipeline network system, the point of attainment of dew point temperature must be ascertained d to enable installation of drain plugs for periodical removal of the liquid fallout and the condensates so formed. Gas chromatographs, chilled mirror, gas equations of state such as Peng Robinson, Redlich Kwong Soave and PG-SAFT methods had been used at one time or the other for hydrocarbon dew point determination,

even the characterization of the gas stream. This paper is to exploit the fundamental heat flux concept for the determination of dew point temperature of the complex hydrocarbon mixture. This approach would generate comparative analysis with other existing methods of dew point calculation of natural gas stream in pipeline network system by computer simulation of the physical developed models.

#### **III COMPUTER SIMULATION FOR THE PRECISE POSITION OF DEW POINT**

The computational analysis core mathematical models are as outlined:

$$q = \frac{2\pi L(T_1 - T_4)}{\frac{1}{h_{in}} + \ln\left(\frac{r_2}{r_1}\right) / K_{As} + \ln\left(\frac{r_3}{r_2}\right) / K_{cs} + \ln\left(\frac{r_4}{r_3}\right) / K_{mw} + \ln\left[\left(\frac{d}{r_4}\right) + \sqrt{\left(\frac{d}{r_4}\right)^2 - 1}\right] / K_s}$$
(1)

$$h_{in} = 0.0225 \frac{K_f}{2r_1} \left[ \frac{2Vr_1}{v_f} \right]^{0.8} \left[ \frac{C_{Pf} v_f \rho_f}{K_f} \right]^{0.4}$$
(2)

$$q = \frac{2\pi V t (T_1 - T_4)}{\frac{1}{h_{in}} + \ln\left(\frac{r_2}{r_1}\right) / K_{As} + \ln\left(\frac{r_3}{r_2}\right) / K_{cs} + \ln\left(\frac{r_4}{r_3}\right) / K_{mw} + \ln\left[\left(\frac{d}{r_4}\right) + \sqrt{\left(\frac{d}{r_4}\right)^2 - 1}\right] / K_s$$

$$q_R = \dot{m}_L C_{PL} (T_d - T_R) + \dot{m}_G C_{PG} (T_d - T_R)$$

$$T_d = T_R + \frac{q_R}{\dot{m}_L C_{PL} + \dot{m}_G C_{PG}}$$
(3)

## **Computational Algorithm Input Parameters**

Inner radius of corrosion coating,  $r_1=0.4532m$ . Inner pipe radius,  $r_2=0.4572m$ Outer pipe radius,  $r_3=0.4682m$ . Outer radius of insulation,  $r_4=0.50722m$ . Thickness of corrosion coating, th=0.004m Burial depth to the pipe centerline, d=1.4572m. Thermal conductivity of carbon steel, K<sub>CS</sub>=0.1569W/mK. Thermal conductivity of mineral wool block,  $K_{MW}=130\times10^{-3}W/mK$ Internal flow conductivity,  $K_f=0.123W/mK$ . August, 2008 and January 2006 production data(ElfTotal Petroleum Nigeria Limited): Specific heat capacity of liquid and condensate, C<sub>PL</sub>=959.1696J/KgK. Specific heat capacity of gas, C<sub>PG</sub>=968.6694J/KgK. Liquid density,  $\rho_L$ =589.207Kg/m3. Gas density,  $\rho_G$ =73.2404Kg/m<sup>3</sup>.Volume fraction of liquid and condensate in the pipeline, R<sub>L</sub>=0.281546. Reference pressure, P<sub>b</sub>=1.01325bar. Reference temperature,  $T_b=281$ K. Average stream temperature,  $T_A=313K$ .

#### A. Simulation Programme for the Precise Position of Dew Point

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% MAIN PROGEAMME FILE
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% DETERMINATION OF THE EXACT POSITION OF DEW POINT TEMPERUTURE

% Initialization

% Internal pipe wall radius, r2(m)

r2=0.4533; % Pipe outer wall radius, r3(m) r3=0.4772; % outer radius of insulation, r4(m) r4=0.5072; % Burial depth to the pipe centre, d(m)d=1.4572 % Kcs--thermal conductivity of carbon steel(W/mK) Kcs=0.1569; % Thermal conductivity of the soil, Ks(W/mK) Ks=1.85; % Kmw--thermal conductivity of mineral wool block((w/mK) Kmw=130E-3; % Thermal conductivity of corrosion coating-Asphalt, Kas(w/mK) Kas=0.74; % Internal flow thermal conductivity, Kf(W/mK) Kf=0.123 % CPL--specific heat capacity of Liquid and Condensates(J/KgK) CPL=959.1696; % CpG--specific heat capacity of Gas(J/KgK) CPG=968.669471; % August 2008 ElfTotal Production Data % GD--gas density(Kg/m3) GD=73.2404; % LD--liquid density(Kg/m3) LD=589.207; % RL--volume fraction of liquid in the pipeline RL=0.280546; % Volume fraction of the gas flowing, VFG VFG=1-RL; % Average internal flow heat capacity, CPA CPA=RL\*CPL+VFG\*CPG; % Density for two phase flow at no slip condition, D2P(Kg/m3) D2P=LD\*(1-VFG)+GD\*VFG; % Mixture flowrate, OM(m3/s) OM=1.8; % Volume fraction of Liquid and Condensates flowing, QL(m3/s) QL=RL\*QM; % Volume fraction of Gas flowing, QG(m3/s) QG=QM-QL; % T--mean or bulk temperature(K) TA=313; % T3==ambient temperature(K) T3=291; % Td--dew point temperature of the mixture(K) Td=67.5; % Average composition of the gas from production line for the month of August 2008(mole fraction) C1A=0.869859;C2A=0.054574;C3A=0.020709;IC4A=0.004517;NC4A=0.006309;IC5A=0.002178;NC5A=0.0 01787;C6A=0.004627;N2A=0.000598;CO2A=0.034843; % Molar mass components of gaseous M11=16;M21=30;M31=44;M4I1=54;M4N1=54;M5I1=72;M5N1=72;M61=86;M71=100;M81=114;M91=128; M101=142;MN21=28;MCO21=44; MAIR=28.97; mixture(PRODUCTION) % Average molecular mass of gaseous in August, 2008MGASA=(C1A\*M11+C2A\*M21+C3A\*M31+IC4A\*M4I1+NC4A\*M4N1+IC5A\*M5I1+NC5A\*M5N1+ C6A\*M61+N2A\*MN21+CO2A\*MCO21); % Specific gravity of the mixture for the month of August, 2008 GAA=MGASA/MAIR;

% To calculate gas absolute viscosity, GV(Pas--Ns/m2)

% Absolute viscosity of the hydrocarbon components, GVHC, is expressed as: GVHC=(8.188E-3-6.15E-3\*(GAA)+(1.709E-5-2.062E-6\*log10(GAA))\*(1.8\*TA+0.27))\*1.02247E-5; % Absolute viscosity of Nitrogen component GVN=(9.59E-3+8.48E-3\*log10(GAA))\*N2A\*1.02247E-5; % Absolute viscosity of carbon dioxide component GVC=(6.24E-3+9.08E-3\*log10(GAA))\*CO2A\*1.02247E-5; % Absolute viscosity of gaseous mixture, GV(Pas-Ns/m2) GV=GVHC+GVN+GVC; % Internal flow kinematic viscosity, KVf(m2/s) KVf = GV/D2P;% L--length of pipeline(m) forL=0:1000:850000 % D--nomonal pipe diameter(m) forD=0.9144:0.9144:0.9144 % r1--inner wall radius of the pipe(m) r1=(D/2)-0.004: % A1--pipe area of crossection A1=pi\*r1^2; % V--average flow velocity(m/s) V=QM/A1;% t=resident time in the pipeline(s) t=L/V;% m1--mass flowrate of gas fraction(Kg/s) m1=GD\*QG; % m2=mass flowrate of liquid fraction(Kg/s) m2=LD\*OL; % Q1--heat lost t0 the environment(J/s) hin=0.0225\*(Kf/(2\*r2))\*((2\*V\*r2)/KVf)^0.8\*(CPA\*KVf\*D2P/Kf)^0.4;  $Q1 = (2*pi*V*t*(TA-T3))/(1/hin+log(r2/r1)/Kas+log(r3/r2)/Kcs+log(r4/r3)/Kmw+log(((d)/(r4))+(((d/r4))^2-r4)/Kas+log(r3/r2)/Kcs+log(r4/r3)/Kmw+log(((d)/(r4))+(((d/r4))^2-r4)/Kas+log(r3/r2)/Kcs+log(r4/r3)/Kmw+log(((d)/(r4))+(((d/r4))^2-r4)/Kas+log(r3/r2)/Kcs+log(r4/r3)/Kmw+log(((d)/(r4))+(((d/r4))^2-r4)/Kas+log(r3/r2)/Kcs+log(r4/r3)/Kmw+log(((d)/(r4))+(((d/r4))^2-r4)/Kas+log(r3/r2)/Kcs+log(r4/r3)/Kmw+log(((d)/(r4))+(((d/r4))^2-r4)/Kas+log(r3/r2)/Kcs+log(r4/r3)/Kmw+log(((d)/(r4))+(((d/r4))^2-r4)/Kas+log(r3/r2)/Kas+log(r3/r2)/Kas+log(r3/r2)/Kas+log(r3/r2)/Kas+log(r3/r2)/Kmw+log((d)/(r4))+(((d/r4))^2-r4)/Kas+log(r3/r2$  $1)^{0.5}/Ks$ ; % Q2--bulk heat of the gas(J/s)Q2=m1\*CPG\*(TA-T3); % Q3--bulk heat of the condensable component(J/s) Q3=m2\*CPL\*(TA-T3); % O4--bulk heat of the mixture(J/s) 04=02+03; % Q5--residual heat(J/s) 05=04-01; % Tdc--dew point of the condensable components(K) Tdc=T3+Q5/(m1\*CPG+m2\*CPL); if Tdc<=Td disp(' Tdc D L') fprintf('%20.7f\n',Tdc,D,L) else disp(' Dew point not yet established at diameter(D) and length(L)') fprintf('%20.7f\n',D,L,Tdc) end end

# IV MATHEMATICAL MODEL COMPUTATIONAL ANALYSIS

The mathematical models are as outlined in equations 1 to 4. The dew point of the hydrocarbon components is calculated using the Elf Total production data for the month of August 2008 and month of January 2006. Employing the chart for equilibrium constants for hydrocarbons at low temperature range, the dew points of the hydrocarbon mixture obtained by trial and error are as in Tables 1 and 2. The dew point of the gaseous mixture as in Tables 1 and 2 is approximately -205.5°C (67.5K)

		Table 1 Elf Pe	troleum Nigeria Ltd Pr month August, 200	oductio )8	n Data for the			
			R?				-	
					Liquid	.1		
Gas C	omposition	(mol fraction, y <sub>i)</sub>	Equilibrium Constant,	Equilibrium Constant, K <sub>i</sub>		Composition(molfraction,x <sub>i</sub> =y <sub>i</sub> /x <sub>i</sub>		
	C1	0.864899	23			0.037604		
	C2	0.054574	2.7			0.020213		
	C3	0.020709	0.58			0.035705		
	IC4	0.004517	0.13			0.034746		
	NC4	0.006309	0.11			0.057355		
	IC5	0.002178	0.035			0.062229		
	NC5	0.001787	0.026			0.068731		
	C6	0.004627	0.006			0.771167		
	Co <sub>2</sub>	0.034843						
	$N_2$	0.000598						
			Liquid and condensate	holdup.	, R <sub>L</sub> =	1.087749	T	
			Gas holdup,	±.				
			R <sub>G</sub> =1					
	Mole f	raction of liquid appr	oximately equal mole fraction	on of gas	an indication			
	of the g	gaseous mixture attair	ning dew point. At RL=RG=	=1087,	the Dew Point			
	temper	ature=-205.5°C(67.51	K).					
		Table 2: Elf P	etroleum Nigeria Ltd P	roducti	on Data for the			
	i	<b>i</b>	month, January, 20	10	Liquid		i.	
Gas Co	omposition(	mol fraction, y <sub>i)</sub>	Equilibrium Constant,	Ki	Composition(	molfraction,x <sub>i</sub> =	y <sub>i</sub> /x	
	C1	0.707881	23			0.030777		
	C2	0.052663	2.7			0.019505		
	C3	0.026519	0.58			0.045722	T	
	IC4	0.005379	0.13			0.041377		
	NC4	0.007883	0.11			0.071664		
	IC5	0.002642	0.035			0.075486		
	NC5	0.002116	0.026			0.081385		
	C6	0.004207	0.006			0.701167		
	Co <sub>2</sub>	0.034843						
	N <sub>2</sub>	0.000598						
			Liquid and condensate	holdup	. <b>R</b> <sub>1</sub> =	1.067082		
			•	Gas ho	ldup. Ro=	1	+	
	Mole fr	action of liquid appro	ximately equal mole fraction	onofga	an indication		+	
	of the g	aseous mixture attain	ing dew point At RI_RG	=1 -0670	)82. the Dew Point		-+	
	temner	ature=-205 5°C(67 5k		1.007			+	
	compere		<del>- / </del>					

Applying the computational algorithm for Elf Total August 2008 and January 20006 production data dew point was attained at a pipe length of 848km and pipe nominal diameter of 36"(0.9144m). The dew point temperature being  $T_d$ =67K. These results are subject to environmental pressure and temperature of 1bar and 291K respectively.

# V MODEL APPLICABILITY

The computer simulated model is applicable to any gas network system to determine the dew point temperature of any gas or complex hydrocarbon mixtures and the exact distance from the upstream or downstream end of the pipeline. The determined dew point temperature is subject to the design and operational conditions of the line.

## VI CONCLUSIONS

A computational algorithm has been developed to enable the determination of dew point temperature of a gas (hydrocarbon gaseous mixtures) and the precise points or distance relative to the inlet or outlet of the gas pipeline network system. The results obtained are remarkable. It is believed this approach will lend itself to optimal performance of our gas pipelines.

#### REFERENCES

- [1].
- Ikoku, Chi. U. (1984): "Natural Gas Production Eng'g", John Wiley and Sons Inc, New York. Guo, B.; Ghalambor, A. and Xu, C. (2005): "A Systematic Approach to Producing Liquid Loading in Gas Wells", paper [2]. Guo, B.; Ghalambor, A. and Xu, C. (2005): SPE94081, presented at the
- [3]. SPE production operations symposium, Oklahoma City, April 17-19. Dojey, A.; Jafferet, C.; Cornot-Gandolph, S.; Jullin, S and Valans, M. (1997):
- [4]. "Natural Gas Production, Processing and Transport", Edition Techrup, Paris, France.
- George D. L., Barajas A. M., Burkey R. C., The need for accurate hydrocarbon dew point determination [J]. Pipeline & Gas [5]. Journal, 2005 September.
- George D. L., Development of Accurate Methods for Predicting Hydrocarbon Dew Point [R]. San [6].
- Antonio: Southwest Research Insitute, 2007. [7].
- Ernest K., Douglas J. P., Hydrocarbon Dew Poit Monitoring of Natural Gas Using Field-Mounted [8].
- [9]. Online Gas Chromatograph {J}, Pipeline and Gas Journal, 2005{7):43.

Dr. Mathew. "Computer Simulation of Precise Dew Point Initiation in Gas Pipelines Network System." International Journal Of Engineering Research And Development, vol. 14, no. 05, 2018, pp. 09-14.